# Quantum Mechanical Study of $\mathrm{C}_{9} \mathrm{H}_{19}-\mathrm{C}_{6} \mathrm{H}_{5}-\mathrm{C}_{6} \mathrm{H}_{5}-\mathrm{CN}$ Cyano Biphenyl molecule 

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#### Abstract

The quantum mechanical calcualtion werer carried out on 4-Alkyl 4'-Cyano Biphenyls ( $\mathrm{C}_{9} \mathrm{H}_{19}-\mathrm{C}_{6} \mathrm{H}_{5}-\mathrm{C}_{6} \mathrm{H}_{5}$-CN). These compounds are very much useful for liquid crystal display devices. The IR spectra and Raman activities and its vibration association were discussed. The atomic charges with multipole moments were also computed and discussed.


Keywolds: Cyano Biphenyl, Liquid Crystals, Mesogen, IR Spectra, Raman Activities.
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## I. INTRODUCTION:

Alkyl and alkoxy cyano biphenyls are higly studied liquid crystals which are suitable for application in electro-optic devices and their properties as liquid crystal devices were well established [1,2]. Dunmur et al. [3] were measured the electric permittivities, refractive indices and densities of the homologous series of alkyl-cyano-biphenyls as a function of temperature in the various phases. Merkel et al. [4] has calculated and analyzed the vibration spectra for cyanobiphenyl liquid crystals using DFT methods. Wu et al. [5] were studied the infra red applications of Perdeuterated cyanobiphenyl liquid crystals and found that it exhibits a much cleaner and reduced infrared absorption. Bernard et al. [6] studied the vibrational spectra of 4-octyloxy, 4'-cyanobiphenyl CN stretching in smectic, nematic, isotropic, and solution phases. Delabre et al. [7] studied the specificities of wetting behaviour of the series of cyanobiphenyl liquid crystals (LCs) on usual substrates, i.e. oxidized silicon wafers, water and glycerol, at both the macroscopic and microscopic scale, in the nematic range of temperature. Paterson et al. [8] synthesized and studied the role of a terminal chain in promoting the twist-bend nematic phase. Wang et al. [9] synthesised and studied the properties of hydroxy tail-terminated cyanobiphenyl liquid crystals.

In the this paper we will discuss about IR as well as Raman activities of 4-Alkyl 4'-Cyano Biphenyls $\left(\mathrm{C}_{9} \mathrm{H}_{19}{ }^{-}\right.$ $\left.\mathrm{C}_{6} \mathrm{H}_{5}-\mathrm{C}_{6} \mathrm{H}_{5}-\mathrm{CN} ; 9 \mathrm{CB}\right)$. The geometry were taken from article by Murty et al. [10]

## COMPUTATIONAL METHOD:

The geometry was optimized using DFT method B3LYP [11,12] using 6-31G** [13, 14] which was found suitable for these type of systems [15] with keeping all atoms free. The analytical frequencies as well as Raman activities were calculated. All calculation were done using Gaussian09 programme suit.[16]

## II. RESULTS:

The optimized geometry of 9CB molecule is shown in figure 1 . The inter ring angle between biphenyl is $37.1^{\circ}$ and inter ring seperation is $1.48 \AA$. The angle between biphenyl and alkyl chain is $88.3^{\circ}$ and seperation is $1.52 \AA$. The cyano group is planar to biphenyl ring and seperation is $1.43 \AA$.


Figure 1: The optimized structure of the 9CB molecule.
Table 1 present the charges as well as multipoles corresponding each atoms of 9CB molecule. Since multipole depends on the coordinates of each atoms therefore coordinates are aslo tabulated here.

Table 1: The charge, coordinates and multipoles corresponding each atoms of 9CB molecule.

| Sr. <br> No. | Atom | Coordinates |  |  | Charge | Multipole (au) |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | X | Y | Z |  |  |  |  |
| 1 | C | -7.28584 | -0.80234 | 0.15476 | -0.41472 | 2.163058 | 0.691982 | -0.11544 |
| 2 | C | -6.38667 | -1.44759 | -0.71108 | -0.50104 | -0.21483 | -0.37504 | -0.23279 |
| 3 | C | -5.0635 | -1.02852 | -0.77703 | -0.11962 | -0.38434 | -0.08979 | 0.014535 |
| 4 | C | -4.5959 | 0.04076 | 0.00895 | 0.219482 | 0.220105 | -0.00987 | -0.08415 |
| 5 | C | -5.50887 | 0.67735 | 0.86973 | -0.08879 | -0.11159 | -0.08467 | -0.02069 |
| 6 | C | -6.83393 | 0.26636 | 0.9474 | -0.38713 | -0.31934 | 0.335216 | 0.399678 |
| 7 | C | -3.18298 | 0.48371 | -0.06798 | 0.047686 | -0.04294 | -0.02514 | 0.052863 |
| 8 | C | -2.49574 | 0.91927 | 1.07741 | -0.22512 | 0.208283 | 0.123707 | 0.030561 |
| 9 | C | -1.17016 | 1.3391 | 1.00067 | -0.08564 | 0.100522 | -0.14033 | -0.81293 |
| 10 | C | -0.47823 | 1.34461 | -0.21811 | -0.21578 | 0.214147 | -0.00997 | -0.56291 |
| 11 | C | -1.16484 | 0.91069 | -1.36052 | -0.3303 | 0.004584 | 0.010291 | -0.08672 |
| 12 | C | -2.48937 | 0.48682 | -1.28998 | -0.1582 | 0.310566 | 0.104008 | 0.154252 |
| 13 | C | 0.97097 | 1.7706 | -0.29216 | 0.34409 | 0.056441 | -0.21283 | 0.092493 |
| 14 | C | 1.95936 | 0.60076 | -0.10623 | 0.072314 | -0.02917 | -0.35917 | -0.10706 |
| 15 | C | 3.42703 | 1.03913 | -0.17753 | 0.576164 | -0.03395 | 0.129852 | 0.065898 |
| 16 | C | 4.4207 | -0.11506 | 0.00653 | 0.367397 | 0.016558 | -0.22406 | -0.07934 |
| 17 | C | 5.88816 | 0.32684 | -0.06053 | 0.448017 | 0.067063 | 0.159798 | 0.025659 |
| 18 | C | 6.88614 | -0.8233 | 0.12471 | 0.441548 | 0.044586 | -0.17236 | -0.03606 |
| 19 | C | 8.35254 | -0.37821 | 0.06075 | 0.368786 | 0.152941 | 0.194639 | 0.011574 |
| 20 | C | 9.35283 | -1.52603 | 0.24802 | 0.422943 | 0.040173 | -0.09896 | 0.000419 |
| 21 | C | 10.8145 | -1.0715 | 0.18519 | 0.203792 | 0.15885 | 0.043216 | 0.132542 |
| 22 | C | -8.6515 | -1.23012 | 0.22872 | 0.43628 | 2.397011 | 0.76076 | -0.12408 |
| 23 | N | -9.76053 | -1.57736 | 0.28886 | 0.427409 | 0.721226 | 0.218077 | -0.03794 |
| 24 | H | -6.7302 | -2.279 | -1.31762 | 0.090106 | -0.01444 | -0.01722 | -0.01298 |
| 25 | H | -4.37292 | -1.55349 | -1.42927 | 0.184311 | -0.00971 | 0.013136 | 0.020826 |
| 26 | H | -5.17968 | 1.5221 | 1.46631 | 0.137122 | -0.00099 | 0.004245 | 0.007679 |
| 27 | H | -7.52786 | 0.77279 | 1.6101 | 0.017335 | -0.05996 | 0.024739 | 0.042051 |
| 28 | H | -2.99666 | 0.90659 | 2.04107 | 0.190506 | 0.048359 | 0.000917 | -0.0351 |
| 29 | H | -0.6615 | 1.66654 | 1.90434 | 0.320654 | -0.10671 | -0.04374 | -0.06248 |

Quantum Mechanical Study of $\mathrm{C}_{9} \mathrm{H}_{19}-\mathrm{C}_{6} \mathrm{H}_{5}-\mathrm{C}_{6} \mathrm{H}_{5}-\mathrm{CN}$ Cyano Biphenyl molecule

| 30 | H | -0.65829 | 0.91855 | -2.32275 | 0.091479 | -0.00749 | -0.01415 | -0.04343 |
| ---: | :--- | ---: | ---: | ---: | ---: | ---: | ---: | ---: |
| 31 | H | -3.00275 | 0.18698 | -2.19895 | 0.218911 | 0.021614 | 0.012645 | 0.049773 |
| 32 | H | 1.16338 | 2.24901 | -1.26089 | -0.07481 | 0.014616 | 0.049808 | -0.10014 |
| 33 | H | 1.16945 | 2.53028 | 0.47439 | -0.02283 | -0.01217 | 0.037203 | 0.045759 |
| 34 | H | 1.76235 | -0.16085 | -0.87262 | -0.18355 | -0.03212 | -0.1146 | -0.11754 |
| 35 | H | 1.76409 | 0.11617 | 0.85976 | -0.11151 | -0.01993 | -0.04676 | 0.128001 |
| 36 | H | 3.61331 | 1.52824 | -1.14444 | -0.20531 | 0.03267 | 0.081659 | -0.16491 |
| 37 | H | 3.61417 | 1.80497 | 0.58888 | -0.22231 | 0.031661 | 0.129028 | 0.13119 |
| 38 | H | 4.23276 | -0.60539 | 0.97235 | -0.17949 | -0.0251 | -0.07481 | 0.150821 |
| 39 | H | 4.23578 | -0.8797 | -0.76141 | -0.23067 | -0.03566 | -0.1404 | -0.13508 |
| 40 | H | 6.07525 | 0.81748 | -1.02649 | -0.20819 | 0.031473 | 0.083667 | -0.16556 |
| 41 | H | 6.07101 | 1.09263 | 0.70695 | -0.2148 | 0.0281 | 0.128909 | 0.128165 |
| 42 | H | 6.6981 | -1.31498 | 1.08999 | -0.19458 | -0.02826 | -0.07817 | 0.155572 |
| 43 | H | 6.70523 | -1.58848 | -0.64385 | -0.21816 | -0.02982 | -0.13244 | -0.1296 |
| 44 | H | 8.54171 | 0.11272 | -0.90485 | -0.20136 | 0.024967 | 0.079249 | -0.16537 |
| 45 | H | 8.53337 | 0.38787 | 0.82863 | -0.20597 | 0.022361 | 0.124446 | 0.124902 |
| 46 | H | 9.16328 | -2.01682 | 1.21251 | -0.18056 | -0.02925 | -0.07851 | 0.148268 |
| 47 | H | 9.17382 | -2.29076 | -0.52023 | -0.16411 | -0.01312 | -0.10628 | -0.10909 |
| 48 | H | 11.5022 | -1.9123 | 0.32345 | -0.10087 | 0.087198 | -0.11582 | 0.004587 |
| 49 | H | 11.04495 | -0.60941 | -0.78163 | -0.05268 | 0.019451 | 0.036257 | -0.09783 |
| 50 | H | 11.03326 | -0.33169 | 0.96382 | -0.12823 | 0.030914 | 0.106031 | 0.102167 |

Various energies components with zero point corrections of 9 CB molecule is tabulated in Table 2.
Table 2: Energies Components such as electronic, thermal and Free energies of 9CB molecules.

| Energies Components | Hartree |
| :--- | :---: |
| Sum of electronic and zero-point Energies | -908.962125 |
| Sum of electronic and thermal Energies | -908.938789 |
| Sum of electronic and thermal Enthalpies | -908.937845 |
| Sum of electronic and thermal Free Energies | -909.01957 |

Table 3 presents dipole monent, exact polarizability, approx. polarizability and hyperpolarizability of 9CB molecules.

Table 3: Dipole monent, exact polarizability, approx. polarizability and hyperpolarizability of 9CB molecules.

| Dipole Monent | 5.9993 debye |
| :--- | :---: |
| Exact Polarizability | 186.209 |
| Approx Polarizability | 325.194 |
| Hyperpolarizability | $-1.80352607 \mathrm{D}+01$ |

The IR spectra of 9CB molecule is shown in figure 2 . From figure 2 it is visual that there are several peak and the highest peak (IR intensity) is at $3074.8659 \mathrm{~cm}^{-1}$. Second peak is at $3086.036 \mathrm{~cm}^{-1}$. The next peak is at $3106.3828 \mathrm{~cm}^{-1}$. All these frequencies are associated with twisting of alkyl chain with respect to biphenyl plane. Another peak hieght is at $2341.886 \mathrm{~cm}^{-1}$. This is associated with bond streching of CN group. Next peak is at $1659.8402 \mathrm{~cm}^{-1}$. This is associated with twisting of phenyl group attached with CN group. Next peak is at $1539.8855 \mathrm{~cm}^{-1}$. This frequency is assoicated with streching of inter bond separation of phenyl ring.


Figure 2: IR Spectra of 9CB molecule.
Figure 3 represent Raman activities of 9CB molecule. There are various peaks and the highest Raman activity is at $1659.8402 \mathrm{~cm}^{-1}$. The frequency is associated twisting of phenyl ring as well as streching of phenyl and CN bond.


Figure 3: Raman activitty of 9CB molecule.

## III. CONCLUSION:

Electronic structure analysis of on 4-Alkyl 4'-Cyano Biphenyls ( $\left.\mathrm{C}_{9} \mathrm{H}_{19}-\mathrm{C}_{6} \mathrm{H}_{5}-\mathrm{C}_{6} \mathrm{H}_{5}-\mathrm{CN}\right)$ molecule is carried out using DFT methods. The IR spectra and Raman activities were explained.

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